SECTION 1 describing qualities of materials

A All materials (glass, wood, rubber, steel, etc) have various properties. What words are used to describe these properties?

these properties:			
Rubber is Rubber is a	material.	13	
Glass is	_ material.	िंदी	
			≻ 1
			\$

Here are some materials and some properties. Make statements to describe the properties of the materials. For example, Steel is strong or Steel is a strong material.

material.		
Materials	Properties	
glass rubber steel polythene wood wool paper porcelain	stiff flimsy strong weak resilient tough brittle flexible	elastic rigid pliable soft hard fragile

Notice that many materials can be described by more than one property. For example, Steel is strong and rigid. We therefore say: Steel is a strong rigid material.

Make similar sentences about other materials, describing them with more than one property if possible.

B We often want to **modify** statements about the properties of materials. For example,

Glass is extremely brittle.
Polythene is very resilient.
Wood is fairly strong.
Rubber is quite tough.
Paper is not very strong.

We can therefore also say:

Glass is an extremely brittle material. Polythene is a very resilient material. Wood is a fairly strong material.

But notice that we must say:

Rubber is quite a tough material.

Paper is not a very strong material.

Make statements about various materials again, but this time modify the statements with the following words: extremely, very, fairly, quite, not very.

C Ask and answer questions about the properties of various materials, using the **modifiers** above. For example,

Is glass very resilient? No, it isn't, it's very brittle.

Is wool an extremely rigid material? No, it isn't, it's a very soft pliable material.

Isn't paper very strong? No, it isn't, it's quite weak.

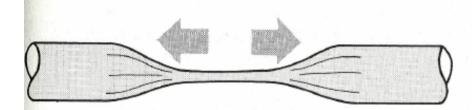
Use the following notes to help you ask the questions:

al			1/1-2247
1	glass/resilient?	4	wool/hard?
! .	glass/rounderittle?	5	paper/strong?
	polythene/brittle?		wood/soft?
3	rubber/rigid?	Ю	WOOd/sore:

- 7 paper/tough?
- 8 steel/weak?
- 9 wool/rigid?
- 10 rubber/brittle?

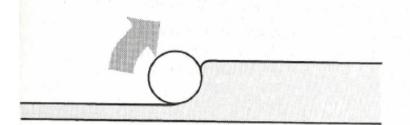
D What words are used to describe the following properties?

A material which can be easily pulled out, or stretched into a long wire or strand, is said to be

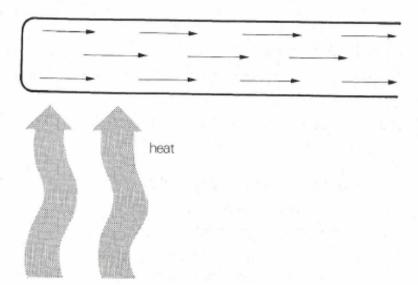


Materials which are used for wires, such as copper and aluminium, must therefore have this property.

A material which can be easily deformed by hammering or rolling is said to be _____



When a substance allows heat or electricity to pass along it, it is said to _____ heat or electricity.



Copper (Cu) and aluminium (AI) are ______, but glass and porcelain are _____.

Make sentences from this table.

Copper (Cu)				222-	A 1
Aluminium (Al)	is	а	extremely very	good	insulator.
Lead (Pb)		an	fairly	poor	conductor.
Glass		1 2 1			
Porcelain					

E You know that we can say: Wood is fairly strong and Steel is very strong. If we wish to **compare** steel and wood, we can say:

Steel is _____ wood.

Now look at these comparisons:

Cardboard is quite strong. Paper is not very strong. Cardboard is slightly stronger than paper.

Steel is very strong. Wood is not very strong. Steel is **much stronger** than wood.

or:

Steel is a lot stronger than wood.

Wool is very soft. Wood is not very soft. Wool is considerably softer than wood.

Rubber is very tough. Paper is not very tough. Rubber is far tougher than paper.

But notice what we say with these properties:

Steel is slightly less ductile/slightly more resilient than copper.

Rubber is much more flexible/much less rigid than steel.

or:

Rubber is a lot more flexible/a lot less rigid than steel.

Glass is considerably more brittle/considerably less resilient than wood.

Polythene is far more resilient/far less fragile than glass.

Make statements comparing these materials:

- 1 glass/fragile/steel
- 2 paper/flimsy/wood

- 3 copper/ductile/iron
- 4 rubber/rigid/steel
- 5 cardboard/stiff/paper
- 6 polythene/brittle/material/glass
- 7 iron/malleable/wood
- 8 paper/strong/cardboard
- 9 porcelain/resilient/material/plastic
- 10 wood/hard/cardboard
- 11 copper/good/conductor/lead
- 12 iron/poor/conductor/aluminium

F All the substances described so far are solids. But, of course, we must be able to describe the properties of other substances. Here are some properties of **liquids** and **fluids**:

oily thick viscous thin creamy runny sticky free-flowing

Name some substances which have some of the properties in the list above. For example,

Milk is a free-flowing white liquid.

Some substances are between solid and liquid form. Such substances may be in the following forms:

gel (jelly) (adjective: gelatinous)
cream (adjective: creamy)
paste

Some solids may be found in the following forms:

powder (adjective: powdery)
crystals (adjective: crystalline)
granules (adjective: granular)

filings chips

flakes (adjective: flaky)

shavings

We sometimes describe these further by using fine or coarse. For example,

Refined sugar consists of fine white granules. Fine iron filings are used to show the presence of a magnetic field.

A substance such as sand may be either fine or coarse.

Now use the words above to describe the following substances as fully as possible.

1	jam	6	oil for a motor car
2	toothpaste	7	sand
3	butter	8	instant coffee
4	salt	9	honey
5	glue	10	chalk

SECTION 2 describing colours and appearance

A What are the three colours of light which together form white light? They are printed on the back cover. What colours are made by mixing these three colours? What are the colours of the spectrum called in English? These are also shown on the back cover.

B We can modify our descriptions of colours by saying:

light blue *or* pale blue dark blue *or* deep blue bright yellow dull brown

Describe the colours of some objects in the

classroom. Ask and answer questions using this table. For example,

What colour is the chair? It's dark brown.

Is the floor light green? No, it isn't. It's dark blue.

light dark pale deep bright dull	red blue green yellow orange purple brown grey pink
---	---

C When an object is not exactly one colour, we can add *-ish* to the colour. For example,

red reddish blue bluish yellow yellowish (but: silver silvery)

When an object is between two colours, we often say: reddish-brown, bluish-yellow, greyish-green, etc. (We can also say: lightish blue, darkish grey, etc.) For example,

Copper is a reddish-brown colour.

The sea is a bluish-green (or greenish-blue) colour.

Describe more of the objects in the room, and objects your teacher presents, using these approximate expressions of colour. Ask and answer questions about the colours of things.

What colours are these?

amber mauve bronze turquoise crimson khaki

D Complete these statements.	Substances which have no colour (like water) are
A material which allows light to pass through it is Glass is a	Water is a liquid.
substance.)	
	A white liquid is sometimes said to be or or Carbon dioxide turns lime water
	When an object or a substance is dirty, it is said to be
A material which does not allow light to pass through it is	E As well as colour, objects have different types of surface or appearance. Surfaces can be:
	bright glossy shiny mat/matt dull
	They can also be: smooth grainy rough corrugated uneven pitted coarse abrasive
A material which allows some light to pass through it is Ground glass or 'frosted' glass is	Complete these descriptions together with your teacher. Your teacher will help you with new words
	1 Glass is a solid which usually has a surface. 2 Chalk is a porous solid which has a surface.
	 Some cardboard is to give it extra strength. The inside of a camera has a surface. Mercury is a liquid metal which has a appearance.

Complete this table:

Adjective	Verb	Noun
hot warm cool		
cold	- 7,	

Complete these statements:

1	At normal	temperatures,	iron i	is a	solid.	However.
	when it is		ove 1.			

2	Water is a	liquid	at normal	temperatur	es.	
	However,	when	it is	below	0°C.	it

3	When	water	is	to 100°0		i+	
---	------	-------	----	----------	--	----	--

- 4 When a substance changes from a solid to a liquid, it is said to ______.
- When a substance changes from a liquid to a solid, it is said to ______.
- 6 When a substance changes from a liquid to a gas, it is said to ______.

C We can also form nouns and verbs from the adjectives in this table. Study the table and write out a completed copy.

Adjective	Verb	Noun
weak tough soft hard rough coarse	weaken	weakness

However, not all nouns and verbs can be formed in this way. What are the missing words from this table? (In some cases we cannot form a verb directly from an adjective, for instance with *resilient*. In this case we have to say *make something resilient*.)

Adjective	Verb	Noun
strong resilient brittle flexible elastic pliable smooth rigid ductile malleable		

0	Fill in the missing words in the following passage:
	Hardening. Medium and high carbon steel are not very h, and so they must be h They are h slowly to a h temperature (above 700°C), and then rapidly c (or 'quenched'). Fully h steel is, however, extremely brittle, and has poor shock resistance. The b of h steel can be reduced and the quality of the metal increased by tempering. This process is described below.
	Tempering. The metal is re-h to a comparatively l temperature and again q at a carefully controlled temperature. The colour of the film of oxide on the brightened surface of the h steel gives a good approximate indication of the temperature of the

exercise 1 Read this passage carefully, and then complete the exercise.

Steel is an alloy of iron (Fe) and carbon (C). The carbon content must not be more than 1.7%. Steel is very strong and ductile. When the carbon content is high, the steel is stronger and harder than when it is low, but steel with a high carbon content is less ductile and less resistant to shock. Steel which contains less than 0.1% carbon is known as **dead mild steel**. It is soft and ductile. Steel which contains up to 0.35% carbon is known as **mild steel**. It is harder and less ductile than dead mild steel.

Steel with a carbon content of 0.35% to 0.7% is called **medium carbon steel**. It is harder, stronger and less ductile than mild steel.

Steel with a carbon content of 0.7% to 1.5% is called **high carbon steel**. It is harder and less ductile than low carbon steels.

Now make these notes into sentences. For example,

mild steel/ductile/medium carbon steel
Mild steel is more ductile than medium carbon
steel.

- 1 steel/high carbon content/ductile/steel/low carbon content
- 2 dead mild steel/soft/mild steel
- 3 medium carbon steel/hard/mild steel
- 4 high carbon steel/ductile/dead mild steel/mild steel
- 5 medium carbon steel/ductile/mild steel
- 6 medium carbon steel/hard/high carbon steel
- 7 mild steel/hard/dead mild steel
- 8 high carbon steel/much/hard/dead mild steel
- 9 dead mild steel/far/soft/high carbon steel
- 10 mild steel/considerably/ductile/high carbon steel

exercise 2 Construct statements which give the main properties and appearance of these substances. For example,

Mercury is a bright shiny white liquid metal.

- 1 Silver 7 Sulphur (S)
- Water 8 Milk
- 3 Copper (Cu) 9 Sulphuric acid (H2SO4)
- 4 Iron (Fe) 10 Coal 5 Coffee 11 Sand 6 Glass 12 Sugar.

exercise 3 From the notes, make sentences like this example:

water/heat/100°C/boil When water is heated to 100°C, it boils.

- 1 aluminium/heat/659·70°C/melt
- 2 water/cool/0°C/freeze
- 3 steel/hard/become/brittle
- 4 hardened steel/heat/270°C/turn/purple
- 5 water/heat/above/100°C/vaporize
- 6 ice/heat/0°C/melt
- 7 liquid steel/cool/solidify
- 8 rubber/vulcanize/become/tougher
- 9 copper/heat/become/more ductile
- 10 glass/tough/become/more resilient
- 11 blue copper sulphate crystals/heat/turn/white
- 12 steel/heat/300°C/film of oxide/turn/dark blue

drill 1 Here is a table which compares the properties of aluminium and copper. Use it to help you make statements about the metals in answer to the questions you hear. For example,

What about strength?

Aluminium is less strong than copper.

What about resistance?
Aluminium has a higher resistance than copper.

Aluminium	Copper
Light Fairly strong Good conductor Fairly cheap Fairly low resistance Very high corrosion	Not very light Very strong Very good conductor Not very cheap Very low resistance Fairly high corrosion
resistance Not very easy to solder	resistance Very easy to solder

drill 2 Here is a list of substances, showing which are good conductors of heat and which are poor conductors. Use the notes to help you answer the questions.

GOOD
1
POOR

Is copper a better conductor than lead? Yes, it's far better.

Is brass a better conductor than copper? No, it's much poorer.

1	far	5	slightly
2	much	6	far
3	much	7	much
4	slightly	8	slightly

drill 3 Here is a table showing the colours of metallic elements. Use it to help you answer the questions.

Copper's brown, isn't it?

No, it's not; it's a reddish colour.

Gold's a dull white, isn't it?

No, it's not; it's a bright yellow colour.

Element	Colour	
copper (Cu)	reddish	
mercury (Hg)	silvery-white	
aluminium (AI)	light white	
zinc (Zn)	bluish-white	
sulphur (S)	pale yellow	
gold (Au)	bright yellow	
lead (Pb)	bluish-white	
magnesium (Mg)	silvery-white	
silver (Ag)	bright white	
tin (Sn)	silvery-white	